

Comparison of microcanonical MD and MC simulations for liquid-gas like phase transitions

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November 27, 2014



Introduction:

Molecular dynamics (MD):

- Based on Newton's equations of motion
- We do constant energy simulations with the Velocity Verlet integrator

Monte Carlo (MC):

- Based on ensemble formulation
- Applied methods: MUCA, Metropolis-Hastings

Motivation:

- Recently mesoscopic systems are studied with a moderate number of degrees of freedom
- How comparable are MD and MC in the microcanonical ensemble?

Introduction:

Microcanonical (NVE) ensemble:

- Motivated by Newtons equations of motion → MD should be the native simulation technique
- Temperature is only an observable in the NVE ensemble
- Monte Carlo simulations can also produce NVE ensemble data (MUCA [1], NVE Metropolis [2])

Our aim:

- Reproduce quantitative microcanonical MC result by developing MD analysis techniques
- Implement a multi histogram reweighting method for MD

¹B.A. Berg and T. Neuhaus, Phys. Lett. B 267 (1991) 249; Phys. Rev. Lett. 68 (1992) 9; W. Janke, Int. J. Mod. Phys. C 03 (1992) 1137; Physica A 254 (1998) 164

²J.R. Ray, Phys. Rev. A 44 (1991) 4061

Ensembles:

Microcanonical NVE ensemble:

$$\Gamma = \int_{\mathbf{x}} \int_{\mathbf{p}} \mathbf{d}\mathbf{x}^N \mathbf{d}\mathbf{p}^N \delta(E - (E_{kin}(\mathbf{p}) + E_p(\mathbf{x})))$$

¹R. Lustig, J. Chem. Phys 100 (1994) 3048

²F. Calvo, J.P. Neirrotti, D.L. Freemann and J.D. Doll, J. Chem. Phys 112 (2000) 10350

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Molecular dynamics or NVEPJ ensemble [2]:

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Molecular dynamics or NVEPJ ensemble [2]:

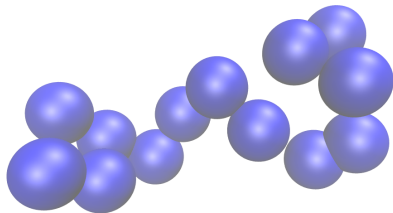
$$\Gamma = \int_{\mathbf{x}} \int_{\mathbf{p}} \mathbf{d}\mathbf{x}^N \mathbf{d}\mathbf{p}^N \delta(E - (E_{kin} + E_p)) \cdot \delta(\mathbf{P} - \sum \mathbf{p}_i) \delta(\mathbf{J} - \sum \mathbf{j}_i)$$

For a high number of degrees of freedom the differences should vanish.

¹R. Lustig, J. Chem. Phys 100 (1994) 3048

²F. Calvo, J.P. Neirrotti, D.L. Freemann and J.D. Doll, J. Chem. Phys 112 (2000) 10350

Single polymer with 13 monomers:



NVE vs. NVEPJ problem:

Number of degrees of freedom $f=33$

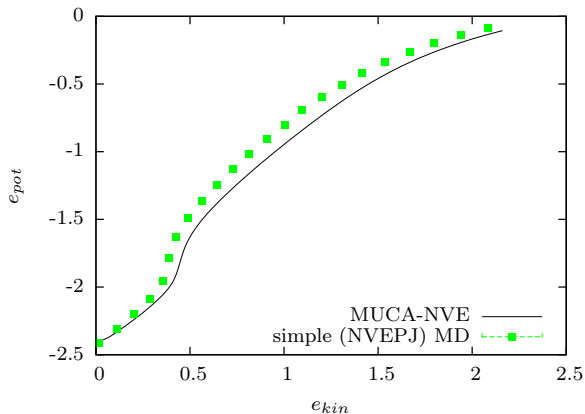


Figure : Kinetic energy-potential energy plot for a 13mer

Reweighting:

Reweighting between NVEPJ and NVE ensemble:

$$\langle O \rangle_{NVE} = \frac{\left\langle O \frac{W_{NVE}(E_p)}{W_{NVEPJ}(E_p, P, J, I)} \right\rangle_{NVEPJ}}{\left\langle \frac{W_{NVE}(E_p)}{W_{NVEPJ}(E_p, P, J, I)} \right\rangle_{NVEPJ}}$$

Reweighted data:

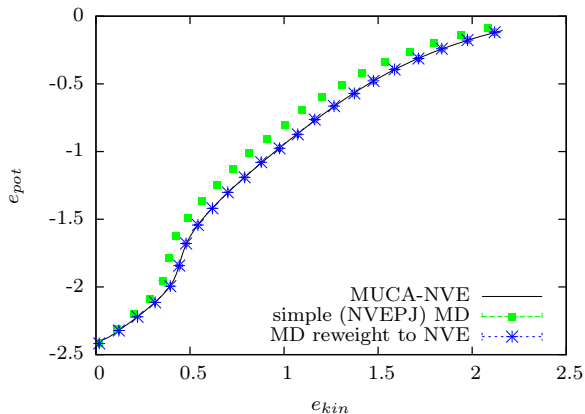
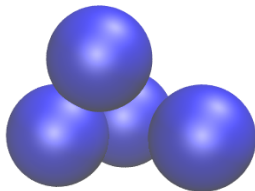


Figure : Kinetic energy-potential energy plot for a 13mer

Lennard Jones gas:

Lennard Jones gas with 4 particles



NVEPJ reweight attempt:

Number of degrees of freedom $f=9$ resp. $f=6$

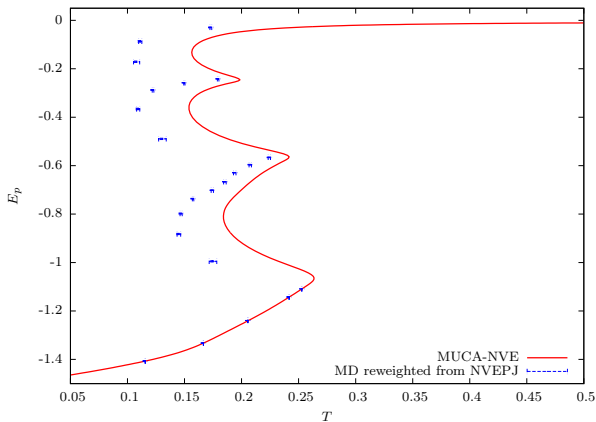


Figure : Temperature-potential energy plot for 4 LJ particles

Reweighting with ensemble determination:

NVEPJ to NVE reweighting for low total energies (condensate):

$$\langle O \rangle_{NVE} = \frac{\left\langle O \frac{W_{NVE}(E_p)}{W_{NVEPJ}(E_p, P, J, I)} \right\rangle_{NVEPJ}}{\left\langle \frac{W_{NVE}(E_p)}{W_{NVEPJ}(E_p, P, J, I)} \right\rangle_{NVEPJ}}$$

NVEP to NVE reweighting for higher total energies (gas):

$$\langle O \rangle_{NVE} = \frac{\left\langle O \frac{W_{NVE}(E_p)}{W_{NVEP}(E_p, P, J, I)} \right\rangle_{NVEP}}{\left\langle \frac{W_{NVE}(E_p)}{W_{NVEP}(E_p, P, J, I)} \right\rangle_{NVEP}}$$

Result:

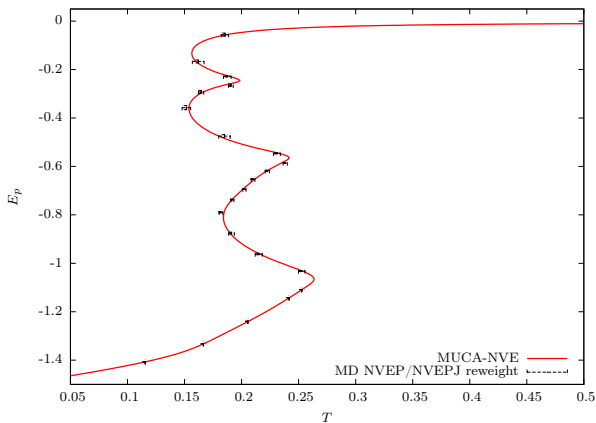


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MD WHAM:

Adapted WHAM procedure:

NVE

NVT [1]

$$W(E_p) = (E_m - E_p)^{\frac{3N-2}{2}} \Theta(E_m - E_p)$$

$$e^{-\frac{E_p}{T_m}}$$

$$\Omega(E_p, Z_1, \dots, Z_M) = C \frac{\sum_m^M H_m(E_p)}{\sum_m^M N_m \frac{1}{Z_m} W_{NVE_m}(E_p)}$$

$$C \frac{\sum_m^M H_m(E_p)}{\sum_m^M N_m \frac{1}{Z_m} W_{NVT_m}(E_p)}$$

$$Z_m = \sum_{E_p=E_{\min}}^{E_m} \Omega(E_p, Z_1, \dots, Z_M) W_{NVE_m}(E_p) \quad \sum_{E_p=E_{\min}} \Omega(E_p, Z_0, \dots, Z_M) e^{-\frac{E_p}{T_m}}$$

¹S. Kumar, J.M. Rosenberg, D. Bouzida, R.H. Swendsen and P.A. Kollman, J. Comput. Chem. 13 (1992) 1011

MD-WHAM result:

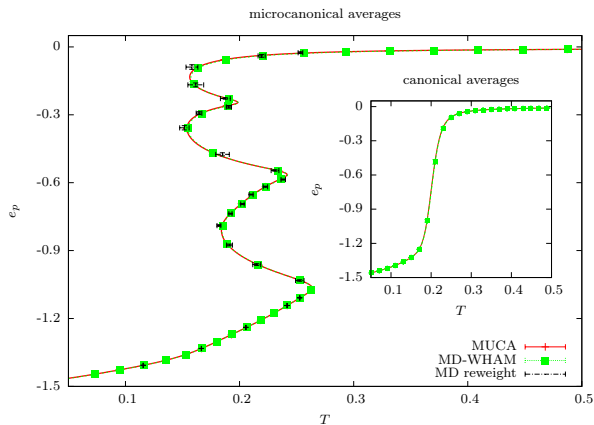
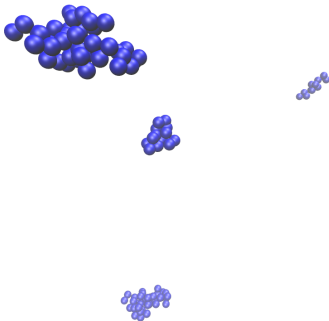


Figure : Temperature-potential energy plot for 4 LJ particles

8 x 13 polymer system:



MD-WHAM result:

Number of degrees of freedom $f=309$ resp. $f=303$

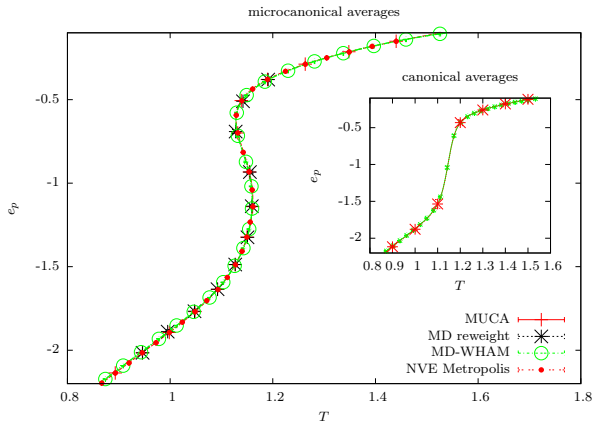


Figure : Temperature-potential energy plot for 8x13mer

Non ergodicity for single MD run:

Differences for separate initial conditions:

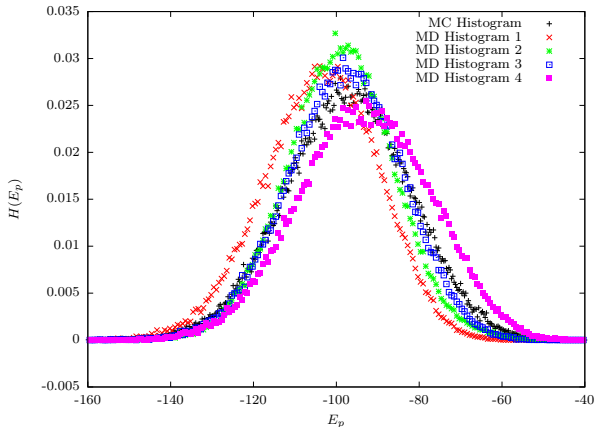


Figure : Histograms at one total energy

Non ergodicity for single MD run:

Merge from ~20 initial conditions:

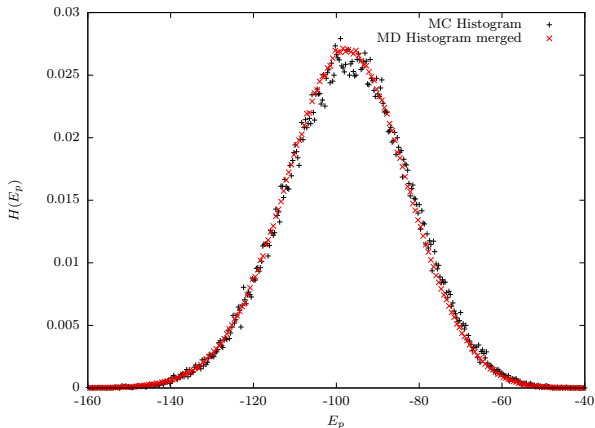


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Comparison to common MD analysis:

Our method:

- time series reweighting

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Common method:

- temperature definition via equipartition theorem:

$$\langle T \rangle = \frac{2}{f} \frac{\langle E_{kin} \rangle}{k_B}$$

- the number of degrees of freedom in this formula takes care of the conservation laws

Comparison to common MD analysis:

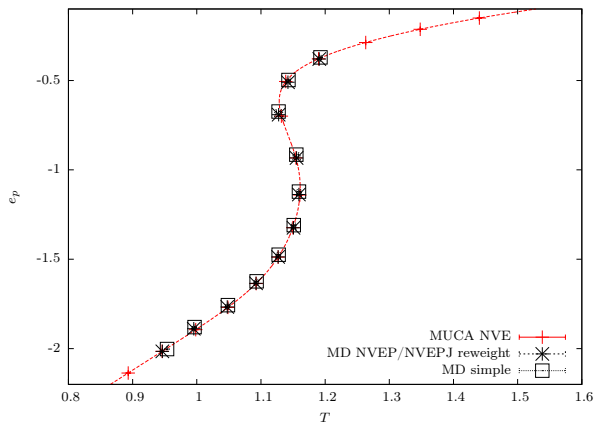


Figure : Temperature-potential energy plot for 8x13mer

Summary:

What we have learned about MD:

- MD samples the NVEPJ or NVEP ensemble for the examples in this work
- MD tends to be non ergodic in a single MD run for more complicated systems
- We can estimate a density of states from MD simulations
- We can estimate NVT averages (losing dynamics) from many MD simulations at different total energies

End

**Thank you for your
attention!**