The stability of foam films: remarks on the concept of disjoining pressure

Introduction

The accepted (DLVO) theory by Derjaguin, Landau, Verwey, Overbeek

Doubts on the key concept of disjoining pressure

Harald Morgner Wilhelm Ostwald Institute for Physical and Theoretical Chemistry Leipzig University, Linnéstrasse 2, D-04103 Leipzig, hmorgner@rz.uni-leipzig.de 1

Introduction Theoretical thin liquid soap films

thin liquid soap films:

reflections, refractions, colors of light

Isaac Newton (1643 – 1727), Robert Hooke (1635 – 1703)

minimal surfaces and their forms

Joseph Antoine Ferdinand Plateau (1801 – 1883) : Statique experimentale et theorique des liquides soumis aux seules forces moleculaires. (Gauthiers-Villars, Paris, 1873)



Tetrahedron

Square

Octahedron



Introduction foam films

Theoretica

Simulation R

Results

froth flotation

selective ore mining

recycling of paper (de-inking)

flotation cell

air bubbles loaded with copper sulfide float on the pulp in a flotation cell



IntroductionTheoreticalSimulationResultsfoam films

The stability of foam films





The official Guinnes world record bubble created 2013 by Megan Colby Parker

IntroductionTheoreticalfoam films



from: Foam Films: Properties and Stability Lecture at Trinity College, Dublin



from: Foam Films: Properties and Stability Lecture at Trinity College, Dublin



positive

Experimental

Simulation

Result

disjoining pressure





C.Ridings and G. Andersson, Rev.Sci. Instr. 81, 2010, 113907 FIG. 2. A) shows the schematic of the film holder viewed from top with the glass capillary fused to the back of the porous disk.

B) is a front view and details the schematic for applying a pressure difference over the film. The filled capillary is connected to a beaker. Both parts of the glass tube are connected via the 90° bend. The difference between the level of solution in the beaker and the height of the film is proportional to the pressure applied.

disjoining pressure

Π=hρg

positive

Theoretical Simulation Results

EOS



EOS

Theoretical Simulation Results

EOS schematic 1000 **Τ0** + -50000 500 + -100000 0 -150000 -500 pressure -200000 chem. pot. pressure - mueA -1000 -250000 -300000 -1500 -350000 -2000 - 400000 -2500 -450000 -500000 -3000 1.00E-04 1.00E-03 1.00E-02 1.00E-01 density [mol/cm^3]





IntroductionTheoreticalSimulationResultliquid film viewed by thermodynamics





mechanical concept (e.g. Bergeron J.Phys.Condens.Matter 11, 1999,R215):

difference in pressure means non-equilibrium of forces (per unit area)

explanation: missing force results from the interaction between both interfaces

 \Rightarrow disjoining pressure must depend on film thickness h, i.e. $\Pi = \Pi(h)$

Experimental Simulation Resu

measurement of disjoining pressure



C. Stubenrauch, J. Schlarmann, R.Strey Phys.Chem.Chem.Phys. 4, 2002, 4504-4513

experimental observation:

disjoining pressure varies with film thickness h, i.e. $\Pi = \Pi(h)$





\Rightarrow

 $\Pi(h)=P_{gas}-P_{liq}$ constant

curvature of interface constant

\Rightarrow

which key quantity controls the disjoining pressure?

answer:

Π depends on chemical potential ¹⁷

physical meaning of disjoining pressure?

Results



Theoretica

Simulation Results

physical meaning of disjoining pressure?

water under N2 pressure full symbol: P_vapor_ref - P_liquid ref



in fact, experiments demonstrate dependence of lamella thickness on chemical potential

disjoining pressure does not correspond to force between interfaces

pressure difference $\Delta P = P_{vap} - P_{liq}$ is counteracted by surface tension of curved interface.

Quantitative agreement with Young-Laplace equation $\Delta P=2 \sigma/R$

the mechanical interpretation of $\Delta P = P_{vap} - P_{liq}$ as "disjoining pressure" jacks justification

Thank you. Critical remarks welcome