



CG vs. AA:
Water

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Comparative atomistic and coarse-grained study of water: What do we lose by coarse-graining?

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Motivation

CG vs. AA:
Water

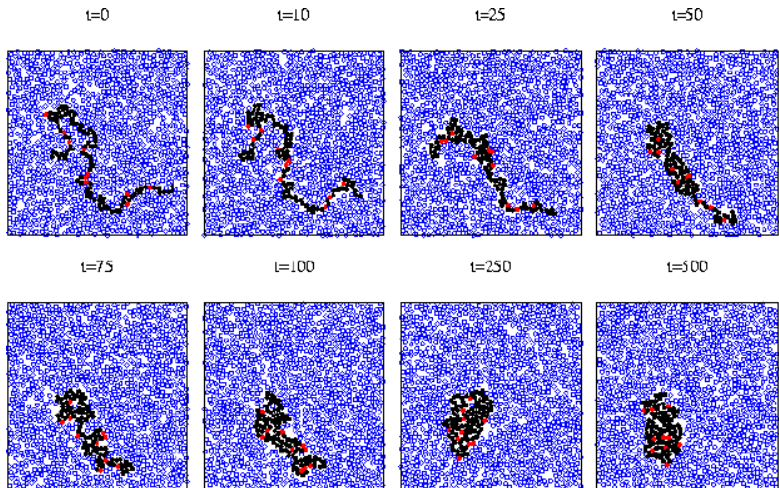
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Conclusion



→ What about the solvent?

¹<http://www.upei.ca/physics/polson/research/research.html>



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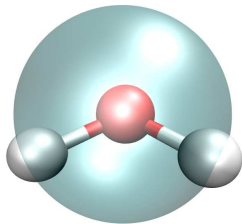
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Replace every solvent molecule by a coarse-grained bead.



Why?

- Save computer time
 - All-Atom** 1 Lennard-Jones
+ 9 Coulomb
 - Coarse-Grained** 1 Tabulated
- Couple MD to continuum
- Understanding general properties

Introduction

Models

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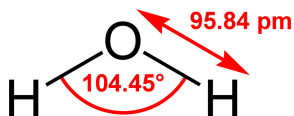
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Pressure-
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Simplest class of water model:

- Rigid geometry
- Lennard Jones interaction between oxygens
- Coulomb (with partial charges) between all atoms

	TIP3P	SPC	SPC/E
r_{OH} [Å]	0.9572	1.0	1.0
$\angle HOH$ [°]	104.52	109.47	109.47
σ [nm]	0.3151	0.3166	0.3166
ϵ [kJ/mol]	0.6364	0.6506	0.6506
q_O [e]	-0.834	-0.820	-0.8476
q_H [e]	+0.417	+0.410	+0.4238

- At ambient conditions, all of them give reasonable results.



Introduction

Coarse-Graining

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Structure based coarse-graining

- Find a pair potential for the coarse-grained system with the equal radial distribution function (RDF, $g(r)$)
- Target RDF $g^{\text{target}}(r)$ is taken from atomistic simulation

Inverse Boltzmann Method²

$$V_{i+1}(r) = V_i(r) + k_B T \ln \left[\frac{g_i(r)}{g^{\text{target}}(r)} \right]$$

with potential of mean force as initial guess

$$V_0(r) = -k_B T \ln g^{\text{target}}(r)$$

Neglects higher (> 2) body effects !

²D. Reith et al., J. Comp. Chem. **24**,1624 (2003).



Introduction

Pressure-Correction

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Pressure will change due to removal of all internal degrees of freedom

Pressure-Correction (PC)

Adding an additional potential of the form

$$\Delta V(r) = A \left(1 - \frac{r}{r_{\text{cut}}} \right),$$

where

$$A_i \approx -(p - p^{\text{target}}) V \left[\frac{2\pi N\rho}{3r_{\text{cut}}} \int_0^{r_{\text{cut}}} dr r^3 g_i(r) \right]^{-1}$$

and at maximum $0.1k_B T$

Results

RDF of SPC/E

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RDF of SPC/E

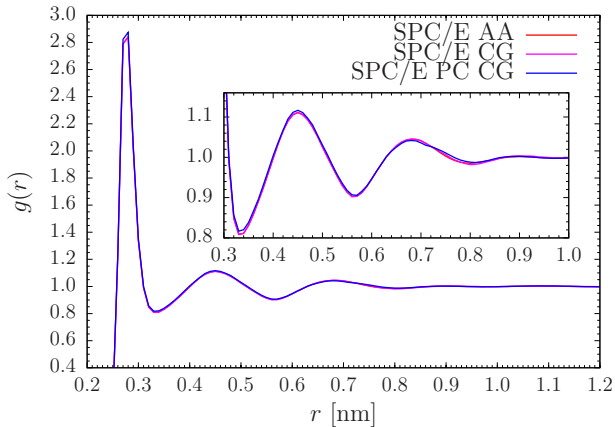
Potentials

Tetrahedral
Packing

Length Scales

Potential of
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→ Pressure correction partly destroys structure

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Potentials

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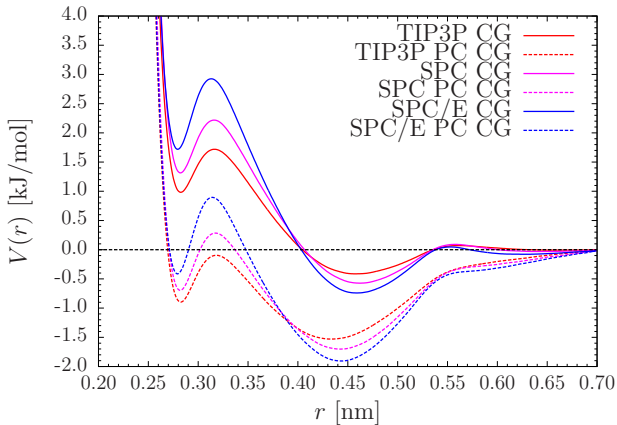
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→ Behavior is governed by two length ratios



Results

Tetrahedral Packing

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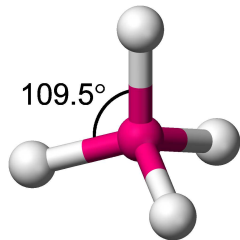
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Water likes to pack in a tetrahedral cluster

- 1 center molecule
- 4 neighbor molecules



Order parameter³

$$q_4 = 1 - C \sqrt{\sum_{i < j} [\theta_{ij} - 109.471^\circ]^2},$$

- $q_4 = 0$ for randomly distributed particles ($\rightarrow C = 1.8165$)
- $q_4 = 1$ for perfect tetrahedral packing

³Other conventions are possible



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Ratio of the wells

- $\gamma = \frac{\text{second well}}{\text{first well}} \approx \frac{2\sqrt{2}}{3}$
- $\frac{\text{neighbor-neighbor distance}}{\text{center-neighbor distance}} = \frac{2\sqrt{2}}{\sqrt{3}}$
- Need for tetrahedral packing

Ratio of maxima

- $\delta = \frac{\text{first well}}{\text{first maxima}} \approx \frac{2}{\sqrt{3}}$
- $\frac{2(\text{center- edge distance})}{\text{center-neighbor distance}} = \frac{2}{\sqrt{3}}$
- Prevents other molecules from entering the cluster

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Length Scales

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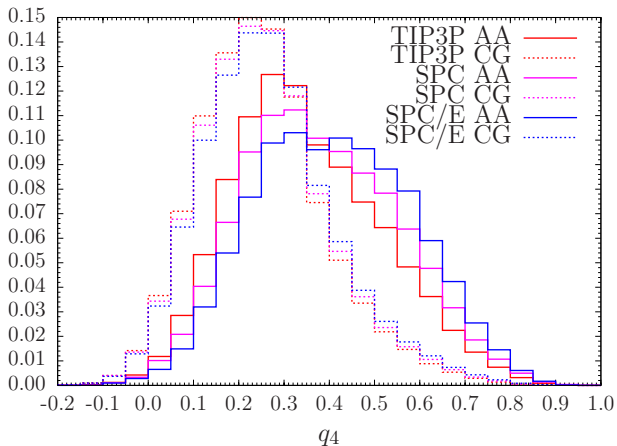
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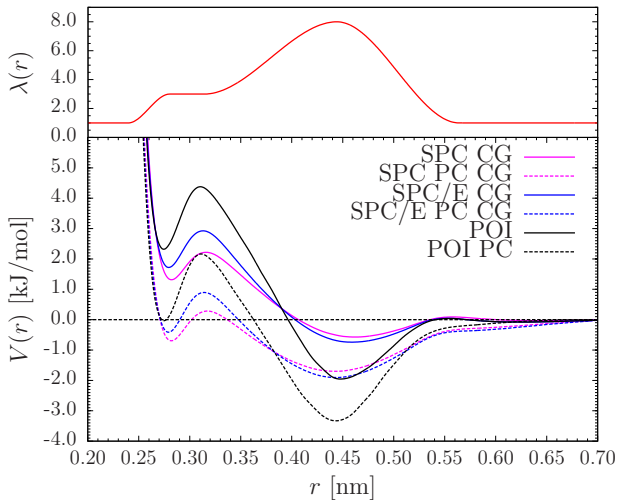


→ Coarse-grained models are less packed

Results

Potential of Extrapolation

Idea: Add more structure by adding $\lambda \cdot (SPC/E - SPC)$



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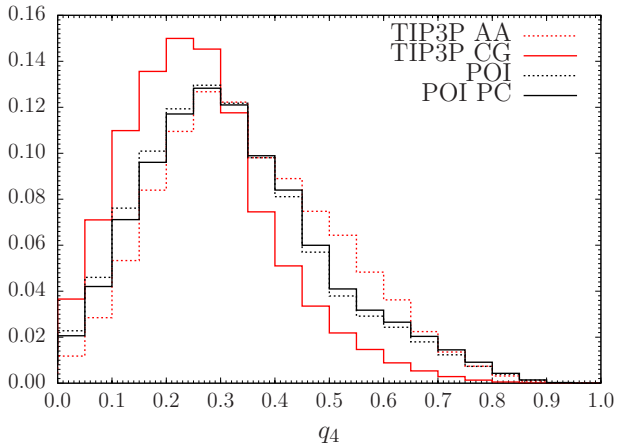
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**Potential of
Extrapolation**

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Potential of Extrapolation

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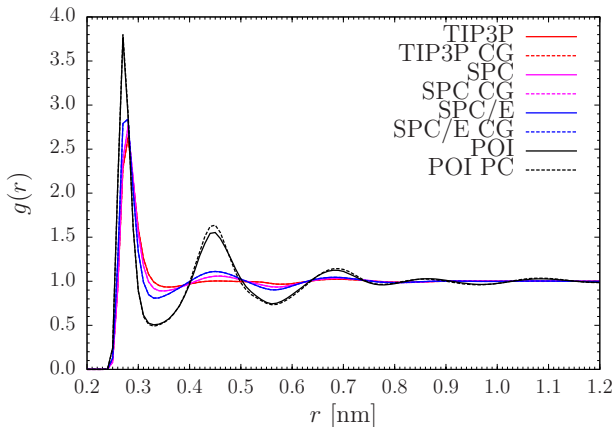
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→ Price to pay: Structure

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Potential of Extrapolation

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Compressibility vs. Pressure

- Want to fit state point (ρ, V, T) and compressibility κ

$$\kappa_T = -\frac{1}{V} \left(\frac{\partial V}{\partial p} \right)_T$$

- Pressure correction destroys compressibility

$$\rho k_B T \kappa_T = 1 + 4\pi\rho \int dr r^2 [g(r) - 1]$$

- Can not fit both at the same time (for this models)

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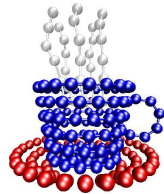
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Issues:

- Compressibility vs. pressure
- Decreased tetrahedral packing
- Structure vs. packing
- Not shown: intrinsic speed-up (from self-diffusion ≈ 5)





The End

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Thank you for your attention !